

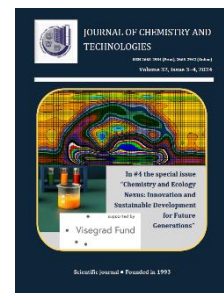


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CRYSTAL GROWTH AND OPTICAL PROPERTIES ANIZOTROPY OF α -NiSO₄×6H₂O

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Abstract

α -NiSO₄×6H₂O single crystal was grown by slow evaporation technique and characterized by X-ray powder diffraction, differential thermal analysis, UV-Vis-NIR, FTIR, and atomic absorption spectroscopy methods. Combination of differential thermal analysis, X-ray powder diffraction and atomic absorption spectroscopy indicates a purity and phase homogeneity of grown α -NiSO₄×6H₂O crystal. The crystal structures of α -NiSO₄×6H₂O was refinement by Rietveld method. The calculated lattice parameters are: $a = 6.8000 \text{ \AA}$, $c = 18.3161 \text{ \AA}$, $Z=4$, space group P4₁2₁2. Four plane-parallel plates with different orientation were prepared. The orientation of the obtained plane-parallel plates was determined by XRD and are as follows: (301), (001), (320) and (204). The optical characteristics – transmittance and optical energy gap of the crystal plates were determined. All the crystal plates are characterized by higher transmittance values in the UVB range than in the VIS and NIR. The transmittance of single crystal plates decreases in the order (204) > (301) > (320) > (001). The band gap E_g values were determined by Tauc equation. Obtained E_g are in the range of 6.43–6.47 eV.

Keywords: growth from solutions; optical transmission spectra; thermal analysis; band gap; X-ray diffraction.

РІСТ КРИСТАЛІВ ТА АНІЗОТРОПІЯ ОПТИЧНИХ ВЛАСТИВОСТЕЙ α -NiSO₄×6H₂O

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Анотація

Монокристали α -NiSO₄×6H₂O було вирощено методом повільного випаровування з водного розчину та охарактеризовано методами порошкової рентгенівської дифракції, диференціального термічного аналізу, оптичної спектроскопії у широкому діапазоні та атомно-абсорбційної спектроскопії. Комбінація результатів диференціального термічного аналізу, порошкової рентгенівської дифракції та атомно-абсорбційної спектроскопії вказує на фазову гомогенність вирощених монокристалів α -NiSO₄×6H₂O. Кристалічну структуру α -NiSO₄×6H₂O було уточнено методом Рітвельда. Розраховані параметри ґратки становлять: $a = 6.8000 \text{ \AA}$, $c = 18.3161 \text{ \AA}$, $Z=4$, просторова група P4₁2₁2. Було підготовлено чотири плоскопаралельні пластини різної орієнтації. Орієнтацію отриманих плоскопаралельних пластинок було визначено методом РФА, і вони відносяться до наступних площин: (301), (001), (320) та (204). Було визначено оптичні характеристики – пропускання та оптична ширина забороненої зони кристалічних пластинок. Усі кристалічні пластини характеризуються вищими значеннями пропускання в УФ-діапазоні, ніж у видимому та ближньому ІЧ. Пропускання монокристалічних пластинок зменшується в ряду (204) > (301) > (320) > (001). Значення ширини забороненої зони E_g були визначені за рівнянням Тауца. Отримані значення E_g знаходяться в діапазоні 6.43–6.47 eV.

Ключові слова: вирощування з розчинів; спектри оптичного пропускання; термічний аналіз; ширина забороненої зони; рентгенівська дифракція.

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Introduction

Nickel sulfate forms a significant number of hydrates $\text{NiSO}_4 \cdot n\text{H}_2\text{O}$ ($n = 1-9$) [1; 2]. However, the most actively studied is a hexahydrate $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$, also known as retgersite [3]. $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ (NSH) is a well-known material which attracts considerable attention due to their wide range of properties: piezoelectric [4], magnetic [5], catalytic [6–10], electrochemical [11] and other [12–14]. However, the main field of application of $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ is a selective UV filter and UV detector [15–18], which is confirmed by the available commercial offers [19; 20]. This is due to a high transmittance in the narrow spectral range of 200–300 nm, which makes them promising working elements in "solar-blind" technology [21; 22]. The optical properties of these crystals have been studied on samples oriented in the (001) plane [23; 24] or are not specified.

$\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ exist in two polymorph forms, namely $\alpha\text{-NiSO}_4 \cdot 6\text{H}_2\text{O}$ (deep emerald-green) and $\beta\text{-NiSO}_4 \cdot 6\text{H}_2\text{O}$ (bluish green) [25]. $\alpha\text{-NSH}$ is a low temperature modification, which are formed by spontaneous nucleation from aqueous solution at temperatures below 55°C [26]. $\alpha\text{-NSH}$ crystallize in tetragonal crystal system, space group (SG) $P4_12_12$ and the unit cell parameters $a = 6.790 \text{ \AA}$, $c = 18.305 \text{ \AA}$ and $Z = 4$ [27]. The $\beta\text{-NSH}$ (mineral nickelhexahydrate) belong to monoclinic system, SG $C2/c$ with parameters $a = 9.880 \text{ \AA}$, $b = 7.228 \text{ \AA}$, $c = 24.1300 \text{ \AA}$, $\beta = 98.38^\circ$ and $Z = 8$ [28].

Mainly, the study of the optical properties of $\alpha\text{-NiSO}_4 \cdot 6\text{H}_2\text{O}$ crystals is carried out on samples

oriented in the plane (001). This is due to a good cleavage on {001} along planes. At the same time, there is no information on the optical properties of $\alpha\text{-NSH}$ crystals oriented relative to different crystallographic planes. The aim of this work is to investigate the anisotropy of the optical properties of oriented $\alpha\text{-NiSO}_4 \cdot 6\text{H}_2\text{O}$ crystal samples.

Experimental

Crystal growth and plates preparation

Single crystals of $\alpha\text{-NiSO}_4 \cdot 6\text{H}_2\text{O}$ were grown from aqueous solution by the slow evaporation method. The starting solution was prepared using crystalline $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ of high purity and bidistilled water. The concentration of the NiSO_4 solution was 0.15 mol/dm^3 , and the pH of the solution was adjusted to 2 with $4 \text{ mol/dm}^3 \text{ H}_2\text{SO}_4$ (controlled by a WTW inoLab Multi 9620 IDS). The NiSO_4 solution was kept in a LabExpert 1021 thermostat at 42 °C until the crystal seed appeared, and further single-crystal growth was carried out at 25 °C. The total growth time was 30 days. As a result, optically transparent and homogeneous emerald-green crystals with a typical habitus were obtained (Fig. 1).

The Ni^{2+} content in single crystals was determined using an Agilent 240AA atomic absorption spectrophotometer. The analysis was performed in the flame of an acetylene-air mixture at a wavelength of 429 nm. It was found that the Ni^{2+} content in the sample under study ($m = 0.267 \text{ g}$), $C_{\text{Exp}} = 587.0 \text{ mg/dm}^3$, is close to the theoretically calculated $C_{\text{Calc}} = 596.5 \text{ mg/dm}^3$.

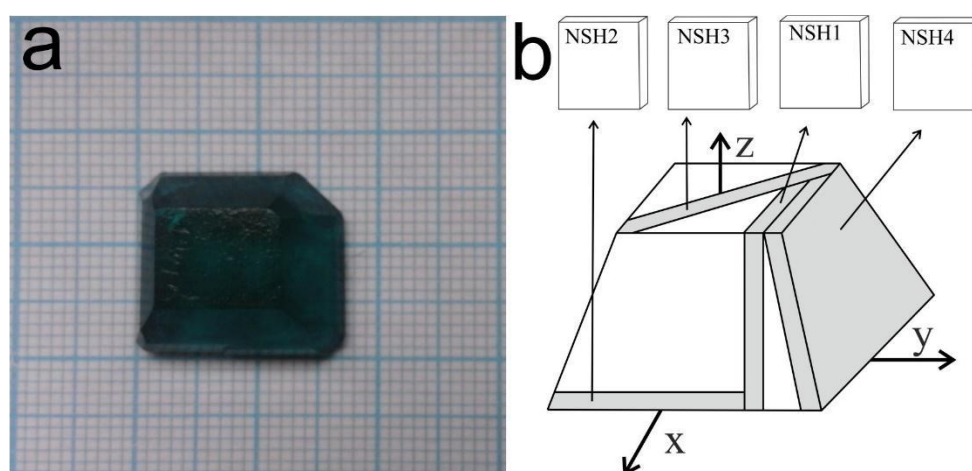


Figure 1. General view of $\alpha\text{-NiSO}_4 \cdot 6\text{H}_2\text{O}$ single crystal (a) and scheme of single crystal plate selection (b)

Four plane-parallel plates marked as NSH1, NSH2, NSH3 and NSH4 with different crystallographic orientations were cut from the grown $\alpha\text{-NiSO}_4 \cdot 6\text{H}_2\text{O}$ single crystal. The surface of

the plates was prepared for further measurements according to the method [29].

Investigation methods

The thermal analysis was carried out in the temperature range of 20–600 °C in atmospheric

air using quartz containers with a heating rate of 12 °C/min. Pre-annealed Al₂O₃ was used as a reference material. The temperature detector is a combined chromel/alumel thermocouple (type K).

The phase analysis of the crystals was studied using an AXRD Benchtop diffractometer equipped with a DECTRIS MYTHEN2 R 1D hybrid detector using Ni filtered CuK_α radiation in the Bragg-Brentano $\theta/2\theta$ mode. The scans were performed in the angle range of 10–120° 2 θ with a dynamic region of interest and an exposure time of 1 s. The phase analysis was performed using the PDAAnalysis (supplied by Proto Manufacturing) program. The lattice parameters were calculated using the EXPO 2014 [30] by Rietveld refinement method [31; 32]. The crystal structure was visualized using VESTA 3.5.4 [33].

The optical parameters of the crystals were studied using inSpect UV-1700 (spectral range

190–1100 nm) and Shimadzu IR Tracer-100 (spectral range 1265–14000 nm) spectrophotometers in transmission mode. For microstructural studies, a METAM-R1 metallographic microscope was used.

Results and discussion

Characterization of bulk α -NiSO₄·6H₂O single crystal

The grown α -NiSO₄·6H₂O single crystal is phase homogeneous, which is confirmed by XRD studies. The experimental and calculated powder patterns (Fig.2) are in good agreement, which is confirmed by the shape of the difference curve. The crystal structures of α -NiSO₄·6H₂O, reported in Ref. [27], was used as structural models for the analysis by the Rietveld refinement method. The calculated parameters of the α -NiSO₄·6H₂O cell are as follows: $a = 6.8000 \text{ \AA}$, $c = 18.3161 \text{ \AA}$, $Z=4$ (Fig. 2).

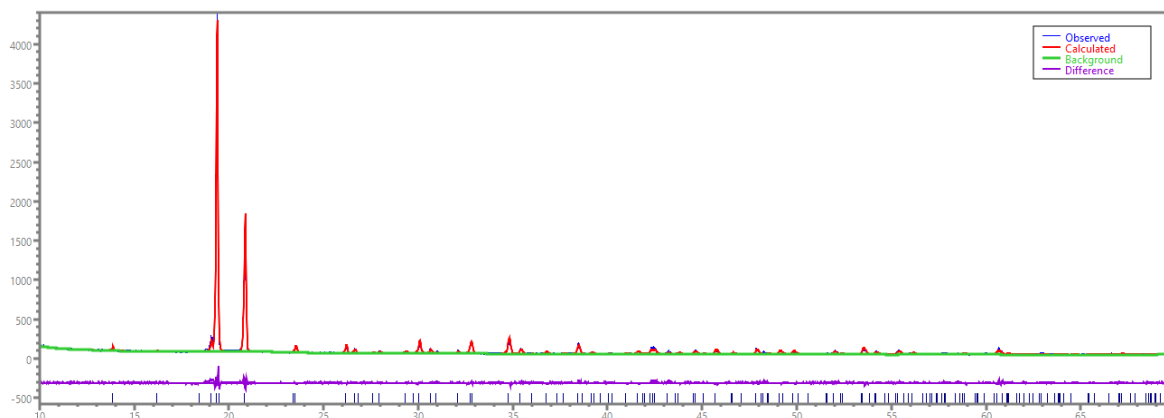


Fig. 2. Experimental (blue line), calculated (red line) and difference (purple) powder patterns of the grown α -NiSO₄·6H₂O

The structure of α -NiSO₄·6H₂O is formed by two types of polyhedra: [SO₄] tetrahedra and [Ni(H₂O)₆] octahedra. All polyhedra [Ni(H₂O)₆] and [SO₄] are isolated from each other, since they do not have any common atoms, but are connected to each other only by hydrogen bonds (Fig. 3). The

[Ni(H₂O)₆] and [SO₄] polyhedra are located at the same height within the cell (parallel to the xy plane) and are surrounded by 4 polyhedra of another type. These planes are equidistant from each other ($d \sim 4.57 \text{ \AA}$). This structure results in good cleavage perpendicular to the z-axis.

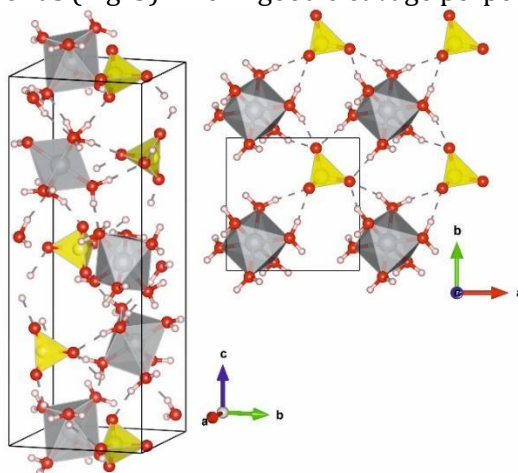


Fig.3. Unit cell of α -NiSO₄·6H₂O

The heating curve of $\alpha\text{-NiSO}_4 \times 6\text{H}_2\text{O}$ is characterized by the presence of three endothermic effects at 105, 126, and 394 °C. The effect at 105 °C is of low intensity and most likely corresponds to the evaporation of water adsorbed on the crystal surface. The effects at 126 °C and 394 °C are accompanied by a significant thermal

effect and correspond to a stepwise dehydration process of $\alpha\text{-NiSO}_4 \times 6\text{H}_2\text{O}$ with the formation of anhydrous yellow NiSO_4 , which was confirmed by XRD. The recorded final mass change $\Delta m = 46.0\%$ is in good agreement with the theoretical one ($\Delta m = 41\%$).

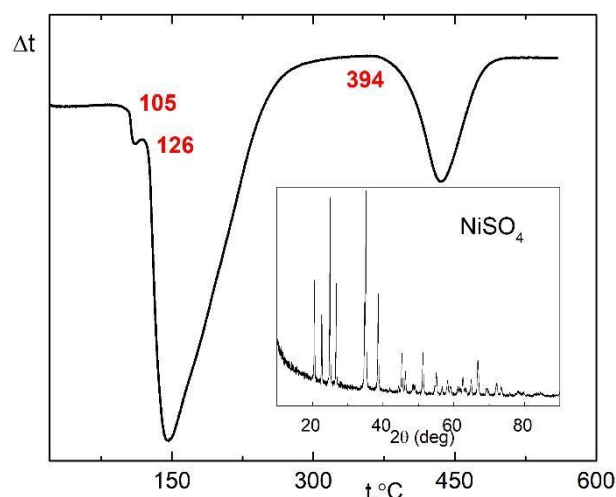


Fig. 4. Heating curve of $\alpha\text{-NiSO}_4 \times 6\text{H}_2\text{O}$. The inset shows the powder pattern of anhydrous NiSO_4

Properties of single crystal plates of $\alpha\text{-NiSO}_4 \times 6\text{H}_2\text{O}$

The orientation of the obtained plane-parallel plates was determined by XRD (Fig. 5). For all samples, the presence of a diffraction peak corresponding to one of the crystallographic

planes is observed: sample NSH1 is oriented along the (301) plane, sample NSH2 – along the (001) plane, sample NSH3 is oriented along the (320) plane, and sample NSH4 is oriented along the (204) plane.

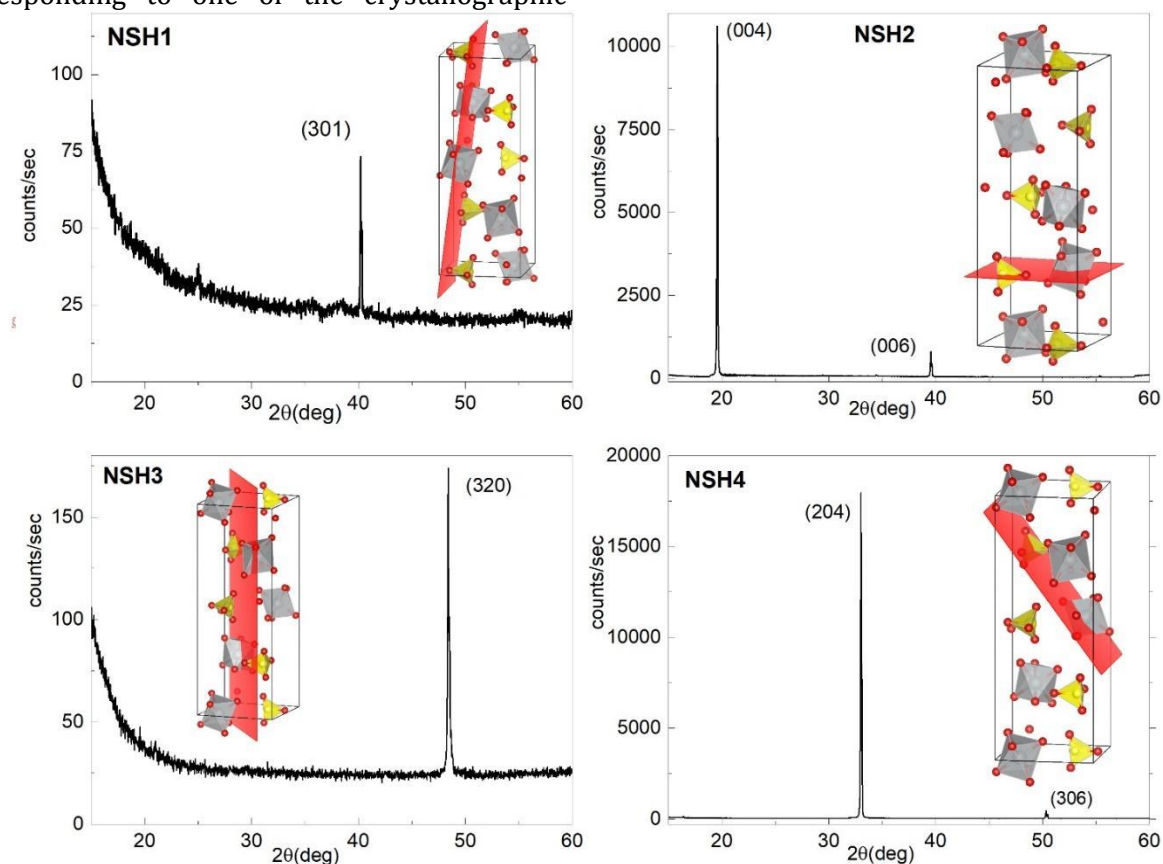


Fig. 5. XRD patterns of the prepared single crystal plates of $\alpha\text{-NiSO}_4 \times 6\text{H}_2\text{O}$

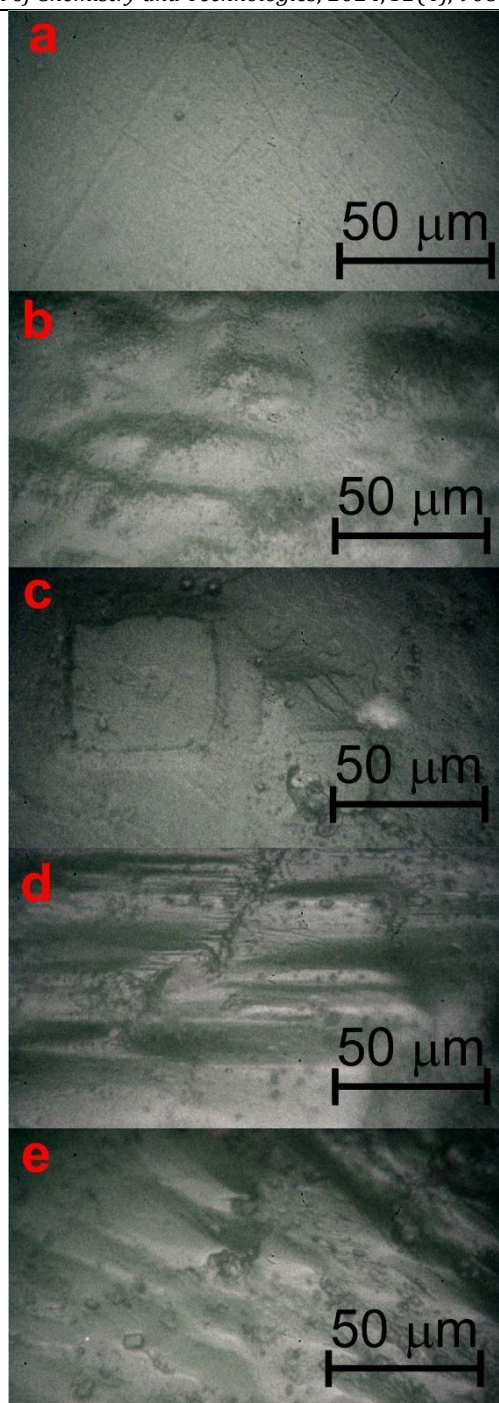


Fig. 6. Surface of untreated plate NSH2 (a) and etched plates NSH1 (b), NSH2 (c), NSH3 (d), NSH4 (e)

To study the optical transmission spectra of α - $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ crystals in the UV-VIS and IR ranges, plane-parallel samples with a thickness of 0.030–0.079 cm were made. The transmittance spectra (UV-VIS-NIR range) of the oriented α - $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ plates (Fig. 7a), in the studied range, contain absorption bands that are typical [29; 33] for the complex cation $[\text{Ni}(\text{H}_2\text{O})_6]^{2+}$ (d^8 configuration) and correspond to optical transitions: ${}^3\text{A}_{2g} \rightarrow {}^3\text{T}_{1g}(\text{P})$ (~ 385 nm), ${}^3\text{A}_{2g} \rightarrow {}^3\text{T}_{1g}(\text{F})$ (~ 645 nm), ${}^3\text{A}_{2g} \rightarrow {}^1\text{E}_g$ (~ 710 nm) and ${}^3\text{A}_{2g} \rightarrow {}^3\text{T}_{2g}$ (~ 1100 nm).

All the crystal plates are characterized by higher transmittance values in the UVB range than in the VIS and NIR, and the maximum transmittance is observed for plates NSH1 and NSH4. The FTIR spectra of the studied α - $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ plates are characterized by the presence of two regions – 1265–2500 nm, characterized by the presence of transmission bands (up to $\sim 40\%$) and a region of strong absorption (>2500 nm), which is most likely due to the presence of the Ni^{2+} ion (Fig.7.b).

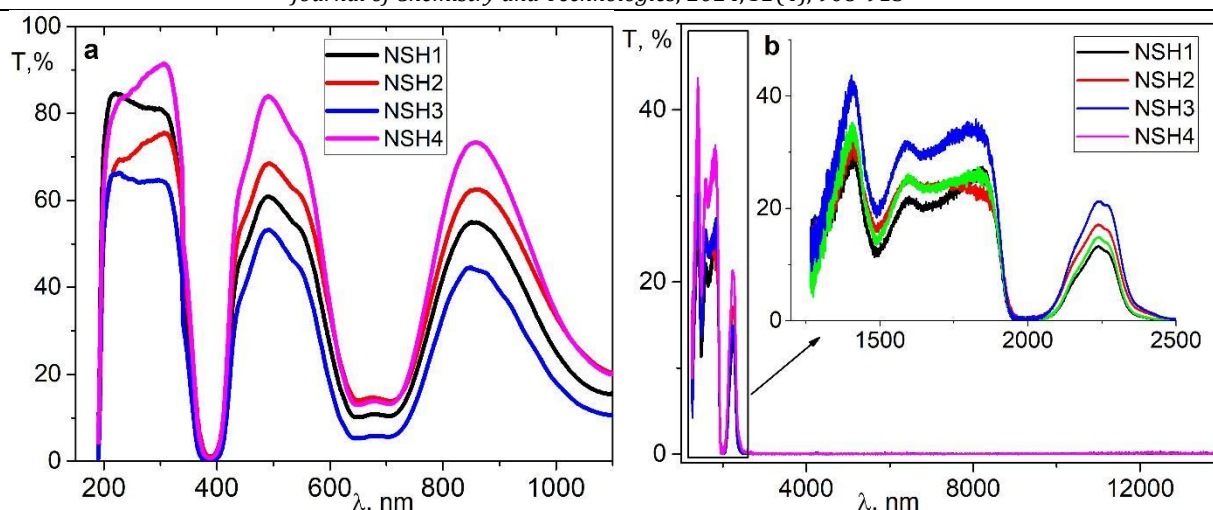


Fig. 7. Transmittance spectra of α -NiSO₄×6H₂O plates in the UV-VIS-NIS (a) and IR (b) ranges

Table 1

Thickness and optical characteristics of α -NiSO₄×6H₂O plates

Sample	Orietation (hkl)	Thickness d, cm	T _{max} , % (at 270 nm)	ln α (at 270 nm)	E _g , eV
NSH1	(301)	0.058	81	1.27	6.44
NSH2	(001)	0.030	73	2.34	6.45
NSH3	(320)	0.079	64	1.72	6.47
NSH4	(204)	0.063	88	0.71	6.43

Since the thickness of the plates is not the same, the absorption coefficient α (for the UV-VIS-NIS range) was calculated. The absorption coefficient α takes into account the thickness of the sample, which allows us to estimate the throughput of samples with different thicknesses. The absorption coefficient α was calculated using the formula [34,35]:

$$\alpha = -\ln(1/T) / d \quad (1)$$

where d - is the thickness of a plate and T - is transmittance.

The analysis of the spectral dependences of $\ln\alpha$ indicates that the highest transmittance is characterized by sample NSH4, and the lowest is characterized by sample NSH2. The transmittance of single crystal plates decreases in the order NSH4 > NSH1 > NSH3 > NSH2. This trend is also clearly visible when comparing the maximum transmittance T and thickness d (Table 1). Thus, the sample NSH2, having the smallest thickness ($\sim 1/2 d$ relative to the others), does not exhibit the highest transmittance.

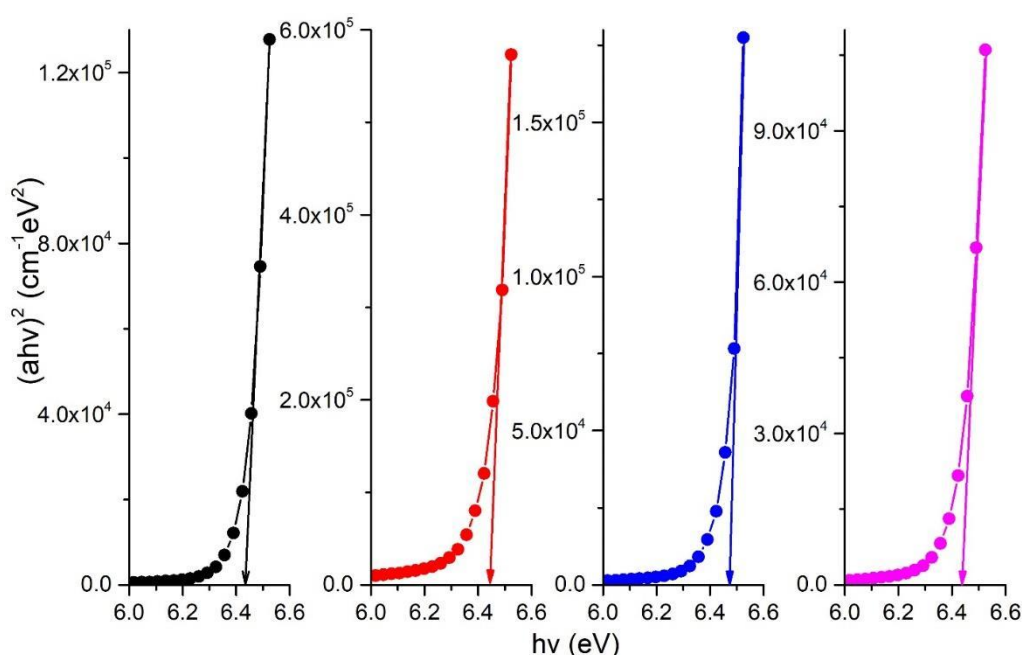


Fig. 8. Tauc's plot for plane-parallel plates of α -NiSO₄×6H₂O

The band gap E_g was determined for the studied α -NiSO₄·6H₂O samples. For this purpose, the Tauc equation [35,36] was used, which combines the absorption coefficient (α) and the frequency of light (ν) with the band gap energy (E_g), and B is the edge width parameter. The parameter m was set as 1/2, which corresponds to a direct allowed transition.

$$ah\nu=B(\alpha h\nu-E_g)^m \quad (2)$$

To determine the band gap E_g value, the corresponding dependencies $(\alpha h\nu)^2$ vs $h\nu$ were constructed, and the corresponding values of E_g were determined by extrapolating the linear section to the intersection of the x-axis. The values of E_g are close (within the determination error) and are in the range of 6.43–6.47 eV.

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Conclusions

α -NiSO₄·6H₂O crystal was successfully grown by slow evaporation technique. Grown crystal was characterized by different methods. XRD results combined with atomic absorption spectroscopy analysis indicate a phase homogeneity of obtained crystals. The optical parameters in the UV-VIS-NIR and IR ranges were studied on plane-parallel α -NiSO₄·6H₂O plates of different orientations. It was found that the samples with orientation (102) have the highest transmittance.

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